

PATENT ABSTRACTS OF JAPAN

(11)Publication number : 09-50810

(43)Date of publication of application : 18.02.1997

(51)Int.Cl. H01M 4/58 H01M 4/02 H01M 10/40

(21)Application number : 07-202286 (71)Applicant : MITSUI TOATSU

(22)Date of filing : 08.08.1995 (72)Inventor : TATSUHIRO KURASAWA
 TAKAO TANAKA
 AKIO YOSIKAWA

CLAIMS

[Claim(s)]

[Claim 1] The electrode active material of the nonaqueous battery using lithium-nickel multiple oxide ($\text{Li}_x\text{Ni}_y\text{N}_z\text{O}_2$, N is elements except Li, Ni and O, $0.8 < x < 1.2$, $0.8 < y+z < 1.2$, and $0 \leq z < 0.2$) coated with lithium-transition-metals M multiple oxide (M is at least one sort of Co, Mn, and Fe, and including a small amount of Ni).

[Claim 2] The manufacturing method;

1. The compound of transition-metals M (M is one or more sorts of Co, Mn, and Fe, and including a small amount of Ni) and lithium compound are dissolved or suspended in a solvent.
2. Lithium-nickel multiple oxide expressed by the general formula $\text{Li}_x\text{Ni}_y\text{N}_z\text{O}_2$ (N is other than Li, Ni, O, $0.8 < x < 1.2$, $0.8 < y+z < 1.2$, and $0 \leq z < 0.2$) is added into above solvent.
3. The slurry is dried and calcinated.

[Claim 3] The nonaqueous electrolyte battery using the electrode active material according to claim 1 for the positive electrode or the negative electrode.

DETAILED DESCRIPTION

[Detailed Description of the Invention]

[The propose of this Invention]

To improve the cycleability of a battery.

AP

[Example 1]

1. Lithium hydroxide monohydrates 83.9g + Nickel hydroxide 185.4g
(atomic-ratio Li/Ni(mole ratio) =1.0)
2. Mill and mix it with ball mill.
3. Dry at 150 degree C for 12 h.
4. The mixture was calcinated at 750 degree C for 5 h in oxygen ambient atmosphere.
5. Mill it for 1 h with ball mill in nitrogen ambient atmosphere.
The average particle size of lithium nickel multiple oxide is 7 um.
6. Ethanol 300g + Lithium nitrate 3.4g + Cobalt nitrate hexahydrate 14.6g
(atomic-ratio Li/Co(mole ratio) =1.0)
7. The mixture + Lithium nickel multiple oxide 92.7g
(atomic-ratio Co/Ni(mole ratio) =0.05)
8. Dry the dispersion with Spray dryer which the temperature of the spray exit is 100 degree C.
9. Calcinate it at 700 degree C for 1 h in oxygen ambient atmosphere.

[Example 2]

1. Lithium nitrate 137.9g + Nickel hydroxide 185.4g
(atomic-ratio Li/Ni(mole ratio) =1.0)
2. Mill and mix it with ball mill.
3. Dry at 150 degree C for 12 h.
4. The mixture was calcinated at 700degree C for 10 in oxygen ambient atmosphere.
5. Mill it for 1 h with ball mill in nitrogen ambient atmosphere.
The average particle size of lithium nickel multiple oxide is 12 um.
6. Ethanol 300g + Lithium hydroxide monohydrate 2.1g + Cobalt nitrate hexahydrate 14.6g
(atomic-ratio Li/Co(mole ratio) =1.0)
7. The mixture + Lithium nickel multiple oxide 92.7g
(atomic-ratio Co/Ni(mole ratio) =0.05)

8. Dry the dispersion with Spray dryer which the temperature of the spray exit is 100 degree C.
9. Calcinate it at 700 degree C for 1 h in oxygen ambient atmosphere.

[Example 3]

1. Lithium hydroxide monohydrate 83.9g + Nickel hydroxide 166.9g + Aluminum hydroxide 15.6g
(atomic-ratio Li/Ni/Al(mole ratio) =1.0/0.9/0.1)
2. Mill and mix it with ball mill.
3. Dry at 150 degree C for 12 h.
4. The mixture was calcinated at 750 degree C for 5 h in oxygen ambient atmosphere.
5. Mill it for 1 h with ball mill in nitrogen ambient atmosphere.
6. Ethanol 300g + Lithium hydroxide monohydrate 2.1g + Cobalt nitrate hexahydrate 14.6g
(atomic-ratio Li/Co(mole ratio) =1.0)
7. The mixture + Lithium nickel multiple oxide 92.7g
(atomic-ratio Co/Ni(mole ratio) =0.05)
8. Dry the dispersion with Spray dryer which the temperature of the spray exit is 100 degree C.
9. Calcinate it at 700 degree C for 1 h in oxygen ambient atmosphere.

[Example 4]

1. Lithium hydroxide monohydrate 83.9g + Nickel hydroxide 166.9g + Cobalt carbonate 137.0g
(atomic-ratio Li/Ni/Co(mole ratio) =1.0/0.9/0.1)
2. Mill and mix it with ball mill.
3. Dry at 150 degree C for 12 h.
4. The mixture was calcinated at 750 degree C for 5 h in oxygen ambient atmosphere.
5. Mill it for 1 h with ball mill in nitrogen ambient atmosphere.

6. Ethanol 300g + Lithium hydroxide monohydrate 2.1g + Cobalt nitrate hexahydrate 14.6g
(atomic-ratio Li/Co(mole ratio) =1.0)
7. The mixture + Lithium nickel multiple oxide 92.7g
(atomic-ratio Co/Ni(mole ratio) =0.05)
8. Dry the dispersion with Spray dryer which the temperature of the spray exit is 100 degree C.
9. Calcinate it at 700 degree C for 1 h in oxygen ambient atmosphere.

[Example 5]

1. Lithium-hydroxide monohydrates 83.9g + Nickel hydroxide 185.4g
(atomic-ratio Li/Ni(mole ratio) =1.0)
2. Mill and mix it with ball mill.
3. Dry at 150 degree C for 12 h.
4. The mixture was calcinated at 750 degree C for 5 h in oxygen ambient atmosphere.
5. Mill it for 1 h with ball mill in nitrogen ambient atmosphere.
6. Ethanol 300g + Lithium nitrate 3.4g + Cobalt nitrate hexahydrate 14.6g
(atomic-ratio Li/Co(mole ratio) =1.0)
7. The mixture + Lithium nickel multiple oxide 92.7g
(atomic-ratio Co/Ni(mole ratio) =0.05)
8. Dry the dispersion with Spray dryer which the temperature of the spray exit is 100 degree C.
9. Calcinate it at 700 degree C for 1 h in oxygen ambient atmosphere.

[Table 1]

	Initial capacity	Capacity at 50 cycles
	初期放電容量 (mAh/g)	充放電 50サイクル後の 放電容量 (mAh/g)
実施例 1	1 6 8	1 5 2
実施例 2	1 7 1	1 5 2
実施例 3	1 6 0	1 4 9
実施例 4	1 5 8	1 5 0
実施例 5	1 5 5	1 4 2
比較例 1	1 4 2	1 2 1
比較例 2	1 3 8	1 2 1
比較例 3	1 1 5	1 0 6
比較例 4	1 3 3	1 1 7

The examples of comparison
Non-coated sample